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Office européen des brevets



(11) Publication number: 0 556 797 A1

(12) EUROPEAN PATENT APPLICATION

(21) Application number: 93102471.5

(51) Int. Cl. 5: B41M 5/00

(22) Date of filing: 17.02.93

(30) Priority: 19.02.92 US 838619

(43) Date of publication of application:
25.08.93 Bulletin 93/34

(84) Designated Contracting States:
AT BE CH DE DK ES FR GB GR IE IT LI LU MC
NL PT SE

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(54) Thermal dye transfer receiving element with backing layer.

(57) A dye-receiving element for thermal dye transfer includes a support having on one side thereof a polymeric dye image-receiving layer and on the other side thereof a backing layer made from a mixture of polyvinyl alcohol as a polymeric binder, submicron colloidal inorganic particles, and polymeric particles of a size larger than the inorganic particles.

EP 0 556 797 A1

This invention relates to dye-receiving elements used in thermal dye transfer, and more particularly to the backing layer of such elements.

In recent years, thermal transfer systems have been developed to obtain prints from pictures which have been generated electronically from a color video camera. According to one way of obtaining such prints, an electronic picture is first subjected to color separation by color filters. The respective color-separated images are then converted into electrical signals. These signals are then operated on to produce cyan, magenta and yellow electrical signals. These signals are then transmitted to a thermal printer. To obtain the print, a cyan, magenta or yellow dye-donor element is placed face-to-face with a dye-receiving element. The two are then inserted between a thermal printing head and a platen roller. A line-type thermal printing head is used to apply heat from the back of the dye-donor sheet. The thermal printing head has many heating elements and is heated up sequentially in response to the cyan, magenta and yellow signals. The process is then repeated for the other two colors. A color hard copy is thus obtained which corresponds to the original picture viewed on a screen. Further details of this process and an apparatus for carrying it out are contained in U.S. Pat. No. 4,621,271.

Dye receiving elements for thermal dye transfer generally include a support bearing on one side thereof a dye image-receiving layer and on the other side thereof a backing layer. As set forth in U.S. Pat. No. 5,011,814, the backing layer material is chosen to (1) provide adequate friction to a thermal printer rubber pick roller to allow for removal of one receiver element at a time from a thermal printer receiver element supply stack, (2) minimize interactions between the front and back surfaces of receiving elements such as dye retransfer from one imaged receiving element to the backing layer of an adjacent receiving element in a stack of imaged elements, and (3) minimize sticking between a dye-donor element and the receiving element backing layer when the receiving element is accidentally inserted into a thermal printer wrong side up.

One backing layer which has found use for dye-receiving elements is a mixture of polyethylene glycol (a double-end hydroxy terminated ethylene oxide polymer) and submicron colloidal silica. This backing layer functions well to minimize interactions between the front and back surfaces of receiving elements and to minimize sticking to a dye-donor element when the receiving element is used wrong side up. This backing layer also provides adequate friction to a rubber pick roller to allow removal of receiving elements from a stack under normal room temperature conditions (20°C, 50% relative humidity). At higher temperatures and relative humidity, e.g. tropical conditions (30°C, 91% relative humidity), however, this backing layer becomes too lubricious and does not allow for effective removal of receiving elements from the supply stack.

U.S. Pat. No. 5,011,814 referred to above discloses a backing layer comprising a mixture of polyethylene oxide (a single-end hydroxy terminated ethylene oxide polymer) and submicron colloidal inorganic particles. By using polyethylene oxide in place of polyethylene glycol in the backing layer mixture, adequate friction is achieved between a rubber pick roller and the backing layer to allow for removal of receiver elements from a supply stack even under high temperature and relative humidity conditions.

U.S. Pat. No. 5,096,875 discloses an improvement over U.S. Pat. No. 5,011,814 wherein polymeric particles of a size larger than the inorganic particles are added to the polyethylene oxide and submicron colloidal inorganic particle containing receiver element backing layer in order to prevent "blocking" or multiple feeding of receiver elements which occasionally results due to too high friction between adjacent receiver elements in the supply stack when using receiver elements having such backing layers.

Polyethylene oxide backing layers, however, have been found to be not as resistant to dye retransfer as would be desirable. It would be desirable to provide a backing layer for a dye-receiving element which would minimize interactions between the front and back surfaces of such elements, minimize sticking to a dye-donor element, provide adequate friction to a thermal printer rubber pick roller to allow for removal of receiver elements from a receiver element supply stack, and control friction between adjacent receiver elements in the supply stack so as to prevent simultaneous multiple feeding of the receiver elements.

These and other objects are achieved in accordance with this invention which comprises a dye-receiving element for thermal dye transfer comprising a support having on one side thereof a polymeric dye image-receiving layer and on the other side thereof a backing layer, wherein the backing layer comprises a mixture of polyvinyl alcohol, submicron colloidal inorganic particles, and polymeric particles of a size larger than the inorganic particles.

In a preferred embodiment of the invention, the backing layer comprises a mixture of 10 to 80 wt.% polyvinyl alcohol as a polymeric binder, 0 to 15 wt.% polyethylene oxide as a polymeric binder, 15 to 80 wt.% submicron colloidal inorganic particles of a size from 0.01 to 0.05 µm, and 1 to 35 wt.% polymeric particles of a size from 1 to 15 µm, the polyvinyl alcohol comprising at least one half of the total amount of polymeric binder by weight.

The process of forming a dye transfer image in a dye-receiving element in accordance with this invention comprises removing an individual dye-receiving element as described above from a supply stack of dye-receiving elements, moving the individual receiving element to a thermal printer printing station and into superposed relationship with a dye-donor element comprising a support having thereon a dye-containing layer so that the dye-containing layer of the donor element faces the dye image-receiving layer of the receiving element, and imagewise heating the dye-donor element thereby transferring a dye image to the individual receiving element. The process of the invention is applicable to any type of thermal printer, such as a resistive head thermal printer, a laser thermal printer, or an ultrasound thermal printer.

In accordance with this invention, adding a polymeric particulate material of the indicated size

decreases the sliding friction between adjacent receiving elements in a supply stack to a greater extent than the picking friction between the backing layer and a rubber pick roller. As a result, blocking or multiple feeding is controlled while adequate picking friction is maintained. Using polyvinyl alcohol in the backing layer mixture results in maintaining adequate friction between the rubber pick roller and the backing layer even under high temperature and relative humidity conditions, while reducing dye retransfer between stacked imaged elements.

The polyvinyl alcohol employed in the invention is preferably essentially fully hydrolyzed and of a molecular weight sufficient to provide a solution viscosity for coating of 10 to 50 cp. Other polymeric binders may be used in combination with the polyvinyl alcohol polymeric binder. Preferably, the total amount of polymeric binder comprises from 10 to 80 wt.% of the backing layer, with at least one-half, preferably at least two-thirds, of the polymeric binder by weight being polyvinyl alcohol.

In one embodiment of the invention, a backing layer polymeric binder combination of polyvinyl alcohol and polyethylene oxide is preferably used for the feature of avoiding sticking of the donor to the receiver backing layer if the receiver is accidentally inserted wrong side up in a thermal printer.

The submicron colloidal inorganic particles preferably comprise from 15 to 80 wt.% of the backing layer mixture of the invention. While any submicron colloidal inorganic particles may be used, the particles preferably are water dispersible and less than 0.1 μm in size, and more preferably from 0.01 to 0.05 μm in size. There may be used, for example, silica, alumina, titanium dioxide, barium sulfate, etc. In a preferred embodiment, silica particles are used.

The polymeric particles may in general comprise any organic polymeric material, and preferably comprise from 1 to 35 wt.%, more preferably 5 to 25 wt.%, of the backing layer mixture. Inorganic particles are in general too hard and are believed to dig into the receiving layer of adjacent receiver elements in a supply stack, preventing such particles from effectively controlling the sliding friction between adjacent receiver elements. Particularly preferred polymeric particles are cross-linked polymers such as polystyrene cross-linked with divinylbenzene, and fluorinated hydrocarbon polymers. The polymeric particles are preferably from 1 μm to 15 μm in size, and particles from 3 μm to 12 μm are particularly preferred.

Additional materials may also be added to the backing layer. For example, improved pencil writability can be obtained, if desired, by the addition of calcined clay. Calcined clays are essentially aluminum silicates that have been heated to remove water of hydration. These materials generally have a particle size of 0.5 to 4 μm , preferably 1 to 2 μm , and may be added at up to 60%, preferably 30-40%, by weight of the backing layer to provide improved writability. Commercially available materials and their average particle size include: Satintone Special (Engelhard Industries), approx 1.2 μm ; Icccap K (Burgess Pigment), approx. 1.0 μm ; Altowhite LL (Georgia Kaolin), approx. 1.8 μm ; and Glomax JDF (Georgia Kaolin), approx. 0.9 μm . Ionic antistat agents may also be added to the backing layer. Surfactants and other conventional coating aids may also be used in the backing layer coating mixture.

The support for the dye-receiving element of the invention may be transparent or opaque, and may be, for example, a polymeric, a synthetic paper, or a cellulosic paper support, or laminates thereof. In a preferred embodiment, a paper support is used for receiving elements for reflective viewing. In a further preferred embodiment, a polymeric layer is present between the paper support and the dye image-receiving layer. For example, there may be employed a polyolefin such as polyethylene or polypropylene. In a further preferred embodiment, white pigments such as titanium dioxide, zinc oxide, etc., may be added to the polymeric layer to provide reflectivity. In addition, a subbing layer may be used over this polymeric layer to improve adhesion to the dye image-receiving layer. In a further preferred embodiment, a polymeric layer such as a polyolefin layer may also be present between the paper support and the backing layer, e.g. in order to prevent curl. Transparent supports may be used for forming images for transparency viewing. For transparencies, the addition of an ionic antistat agent to the backing layer, such as potassium chloride, vanadium pentoxide, or others known in the art, is especially desirable.

The dye image-receiving layer of the receiving elements of the invention may comprise, for example, a polycarbonate, a polyurethane, a polyester, polyvinyl chloride, poly(styrene-co-acrylonitrile), poly-

(caprolactone) or mixtures thereof. The dye image-receiving layer may be present in any amount which is effective for the intended purpose. In general, good results have been obtained at from about 1 to 10 g/m². An overcoat layer may be further coated over the dye-receiving layer, such as described in U.S. Pat. No. 4,775,657.

5 The backing layer of the invention may be present in any amount which is effective for the intended purpose. In general, good results have been obtained at a total coverage of from about 0.1 to 2.5 g/m².

For thermal dye-transfer receivers designed for reflection viewing (such as those having an opaque support), total backing layer coverages of from 0.5 to 2.5 g/m² are preferred. For these backing layers, the total amount of polymeric binder preferably comprises from 10 to 40 wt.% of the backing layer, and a total 10 polymeric binder coverage of 0.1 to 0.4 g/m² is preferred.

For thermal dye-transfer transparency receivers (e.g., those designed for transmission viewing and having a transparent film support), lower total backing layer coverages of from 0.1 to 0.6 g/m² are preferred. Backing layer coverages greater than 0.6 g/m² tend to have too much haze for transparency applications. For these backing layers, the total amount of polymeric binder preferably comprises from 40 to 80 wt.% of 15 the backing layer, and a total polymeric binder coverage of 0.05 to 0.4 g/m² is preferred. Additionally, at least three-fourths of the polymer weight should be polyvinyl alcohol. An especially preferred polymer coverage is polyvinyl alcohol and polyethylene oxide at 0.06 g/m² and 0.02 g/m² respectively. The total polymer coverage is more preferably maintained below 0.25 g/m² to avoid haze.

Conventional dye-donor elements may be used with the dye-receiving element of the invention. Such 20 donor elements generally comprise a support having thereon a dye containing layer. Any dye can be used in the dye-donor employed in the invention provided it is transferable to the dye-receiving layer by the action of heat. Especially good results have been obtained with sublimable dyes. Dye donors applicable for use in the present invention are described, e.g., in U.S. Pat. Nos. 4,916,112, 4,927,803 and 5,023,228.

The dye-donor element employed in certain embodiments of the invention may be used in sheet form 25 or in a continuous roll or ribbon. If a continuous roll or ribbon is employed, it may have only one dye thereon or may have alternating areas of different dyes such as cyan, magenta, yellow, black, etc., as disclosed in U. S. Patent 4,541,830.

In a preferred embodiment of the invention, a dye-donor element is employed which comprises a poly(ethylene terephthalate) support coated with sequential repeating areas of cyan, magenta and yellow dye, 30 and the dye transfer process steps are sequentially performed for each color to obtain a three-color dye transfer image.

Thermal printing heads which can be used to transfer dye from dye-donor elements to the receiving 35 elements of the invention are available commercially. There can be employed, for example, a Fujitsu Thermal Head (FTP-040 MCS001), a TDK Thermal Head F415 HH7-1089 or a Rohm Thermal Head KE 2008-F3. Alternatively, other known sources of energy for thermal dye transfer, such as laser or ultrasound, may be used.

A thermal dye transfer assemblage of the invention comprises a) a dye-donor element as described above, and b) a dye-receiving element as described above, the dye-receiving element being in a superposed relationship with the dye-donor element so that the dye layer of the donor element is in contact 40 with the dye image-receiving layer of the receiving element.

When a three-color image is to be obtained, the above assemblage is formed on three occasions during the time when heat is applied by the thermal printing head. After the first dye is transferred, the elements are peeled apart. A second dye-donor element (or another area of the donor element with a different dye area) is then brought in register with the dye-receiving element and the process repeated. The third color is 45 obtained in the same manner.

The following examples are provided to further illustrate the invention.

Example 1

50 Dye-receivers were prepared by coating the following layers in order on white-reflective supports of titanium dioxide pigmented polyethylene overcoated paper stock:

(1) Subbing layer of poly(acrylonitrile-covinylidene chloride-co acrylic acid) (14:79:7 wt. ratio) (0.08 g/m²) coated from butanone solvent.

(2) Dye-receiving layer of diphenyl phthalate (0.32 g/m²), di-n-butyl phthalate (0.32 g/m²), Fluorad FC-431 55 (a perfluorosulfonamido surfactant of 3M Corp.) (0.01 g/m²), Makrolon 5700 (a bisphenol-A polycarbonate of Bayer AG) (1.6 g/m²), and a linear condensation polymer derived from carbonic acid, bisphenol-A, and diethylene glycol (phenol: glycol mol ratio 50:50, mw ~17,000)(1.6 g/m²) coated from dichloromethane solvent.

(3) Overcoat layer of Fluorad FC-431 (a perfluorosulfonamido surfactant of 3M Corp.) (0.02 g/m²), 510 Silicone Fluid (a partial phenyl substituted polydimethylsiloxane of Dow Corning) (0.02 g/m²) in the linear condensation polymer described above (0.22 g/m²) coated from dichloromethane solvent.

- On the reverse (back) side of these supports a layer of high-density polyethylene (32 g/m²) was extrusion coated. On top of this layer, backing layers of the invention or comparison backing layers were coated from a water and isobutyl alcohol solvent mixture. The backing layers contained polyvinyl alcohol (fully hydrolyzed) from different suppliers, colloidal silica (LUDOX AM alumina modified colloidal silica of duPont) of approximately 0.014 μm diameter, and polystyrene beads crosslinked with m- and p-divinylbenzene of average diameter 12 μm. In some instances polyethylene oxide was added as an additional binder.
- For coating ease, all backing layers also contained Triton X200E (a sulfonated aromatic-aliphatic surfactant of Rohm and Haas) with or without Daxad-30 (sodium polymethacrylate of W. R. Grace Chem. Co.). Phthalic acid monopotassium salt was added as required to maintain the coating pH at 6 to 7.

The following backing layers and controls were prepared:

15 Invention Backing Layer E-1:

Airvol 325 (a fully hydrolyzed polyvinyl alcohol) (Air Products and Chemicals)	0.31 g/m ²
Ludox AM	0.47 g/m ²
Polystyrene beads	0.22 g/m ²
Triton X200E	0.019 g/m ²
Phthalic acid monopotassium salt	0.079 g/m ²

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Invention Backing Layer E-2:

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Elvanol 71-30 (a fully hydrolyzed polyvinyl alcohol) (duPont)	0.24 g/m ²
Ludox AM	0.55 g/m ²
Polystyrene beads	0.22 g/m ²
Triton X200E	0.019 g/m ²
Phthalic acid monopotassium salt	0.079 g/m ²

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Invention Backing Layer E-3:

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Colloids 7190-25 (a fully hydrolyzed polyvinyl alcohol) (Colloids Industries)	0.27 g/m ²
Ludox AM	0.47 g/m ²
Polystyrene beads	0.22 g/m ²
Triton X200E	0.019 g/m ²
Daxad 30	0.019 g/m ²
Phthalic acid monopotassium salt	0.079 g/m ²

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Invention Backing Layer E-4 (as E-3 but different coating coverages):

5	Colloids 7190-25 polyvinyl alcohol Ludox AM Polystyrene beads Triton X200E Daxad 30 Phthalic acid monopotassium salt	0.14 g/m ² 0.65 g/m ² 0.22 g/m ² 0.019 g/m ² 0.019 g/m ² 0.039 g/m ²
10		

Invention Backing Layer E-5 (as E-3 but with polyethylene oxide and adjusted colloidal silica):

15	Colloids 7190-25 polyvinyl alcohol Ludox AM Polystyrene beads Polyox WSRN-10 (a polyethyleneoxide of mw 100,000) (Union Carbide) Triton X200E Daxad 30	0.068 g/m ² 0.65 g/m ² 0.22 g/m ² 0.067 g/m ² 0.019 g/m ² 0.019 g/m ²
20		

25 Invention Backing Layer E-6 (as E-5 but with adjusted coating coverages):

30	Colloids 7190-25 polyvinyl alcohol Ludox AM Polystyrene beads Polyox WSRN-10 Triton X200E Daxad 30	0.14 g/m ² 0.47 g/m ² 0.22 g/m ² 0.14 g/m ² 0.019 g/m ² 0.039 g/m ²
35		

35 A control backing layer based on U.S. Pat. No. 5,011,814 that did not contain polyvinyl alcohol or large particles of polystyrene was also prepared:

40	Control Backing Layer C-1:	
	Ludox AM	0.86 g/m ²
	Polyox WSRN-10	0.13 g/m ²
	Triton X200E	0.019 g/m ²
	Daxad 30	0.089 g/m ²

45 A second control backing layer based on U.S. Pat. No. 5,096,875 containing large particles of polystyrene and small particles of colloidal silica but no polyvinyl alcohol was also prepared:

50	Control Backing Layer C-2:	
	Ludox AM	0.70 g/m ²
	Polystyrene beads	0.22 g/m ²
	Polyox WSRN-10	0.13 g/m ²
	Triton X200E	0.019 g/m ²
	Daxad 30	0.033 g/m ²

To evaluate receiver backing layer to rubber pick roller friction, each dye receiver tested was placed face down (dye image-receiving layer side down) on top of a stack of face down receivers. Two pick rollers (12 mm wide and 28 mm in diameter with an outer 2 mm layer of Kraton G2712X rubber) of a commercial thermal printer (Kodak SV6500 Color Video Printer) were lowered onto the top test receiver so as to come into contact with the backing layer to be tested. The rollers were stalled at a fixed position so that they could not rotate, and supplied a normal force of approximately 4 N (400 g) to the receiver backing layer. A spring type force scale (Chatillon 2 kg x 26 scale) was attached to the test receiver and was used to pull it at a rate of 0.25 cm/sec from the receiver stack. The required pull forces for the various backing layers were measured at high humidity, 90% RH, as the receivers began to slide and are indicated in Table I below. In actual practice, it has been found that pull forces of at least about 6 N (600 g) or more are preferable to ensure good picking reliability.

To evaluate sliding friction between the backing layer of one receiver element and the receiving layer of an adjacent element, a first receiver element was taped to a stationary support with the backing layer facing up. A second receiver element was then placed with its receiving layer face down against the backing layer of the first element. A 1.5 kg steel weight was placed over the two receiver elements, covering an area approximately 10 cm by 12 cm. A cam driven strain gauge was attached to the second (upper) receiver element and advanced about two cm at a rate of 0.25 cm/sec. The maximum pull forces for the various receivers were measured at about 1 sec into the pull and are indicated in Table I below. In actual practice, it has been found that the pull forces of less than about 5 N (500 g) are desirable to prevent blocking or multiple feeding.

To evaluate backside dye-retransfer between the printed receiving layer side of one dye-receiver element and the back of a second dye-receiver an image consisting of a series of individual cyan, magenta, and yellow dye areas was printed by means of a thermal-head as described in Example 2 of U.S. Pat. No. 4,927,803. The transferred Status A reflection densities were approximately 1.0 for each area.

The face of the printed receiver was placed in contact with the backing layer of another unprinted test receiver, placed between two flat metal supports with a 1 kg weight on top, and the assembly was incubated for one week at 50°C, 50% RH. After this time the areas of the test backing that were in contact with the printed areas were read to Status A Red, Green, or Blue reflection density. The background density of an unprinted area was subtracted from each value to obtain the net amount of transferred dye density, which is indicated in Table I below.

Table I

35	Receiver Element	Picking Friction (Newtons)	Sliding Friction (Newtons)	Retransferred Dye Density Status A-Above Background		
				Red	Green	Blue
40	C-1	4.1	6.1	0.05	0.05	0.14
E-1	7.0	3.8	0.03	0.01	0.08	
E-2	7.0	3.2	0.	0.02	0.01	
E-3	7.2	3.5	0.01	0.01	0.	
E-4	7.0	3.0	0.	0.01	0.	
E-5	7.1	3.0	0.	0.01	0.01	
E-6	7.1	3.3	0.	0.01	0.	
50		7.0	3.2	0.	0.01	0.

The data above show that the backing layers of the invention which contain polyvinyl alcohol have excellent high humidity picking friction and sliding friction characteristics compared to the prior art control without large particles.

In addition the invention backing layers with polyvinyl alcohol have significantly less backside dye-retransfer for all three imaged dyes cyan, magenta, and yellow compared to the controls without polyvinyl alcohol.

In a separate experiment backing layers E-5 and E-6, which contain both polyvinyl alcohol and polyethylene oxide, were found to prevent sticking of the donor to the receiver when the receiver was

inserted for printing "wrong side up" in the manner described in the example of U.S. Pat. No. 5,096,875.

Example 2

5 This example is similar to Example 1 but shows the use of the invention backing layers with a transparent polymeric film support for a thermal dye transfer transparency receiver.

Thermal dye-transfer receivers were prepared by coating the following layers in order on a 175 μm thick transparent poly(ethylene terephthalate) support:

(1) Subbing layer of poly(acrylonitrile-covinylidene chloride-co acrylic acid) (14:79:7 wt. ratio) (0.06 g/m²) coated from butanone solvent.

(2) Dye-receiving layer of diphenyl phthalate (0.32 g/m²), di-n-butyl phthalate (0.32 g/m²), Fluorad FC-431 (a perfluorosulfonamido surfactant of 3M Corp.) (0.01 g/m²), Makrolon 5700 (a bisphenol-A polycarbonate of Bayer AG) (1.6 g/m²), and a linear condensation polymer derived from carbonic acid, bisphenol-A, and diethylene glycol (phenol: glycol mol ratio 50:50, mw ~17,000) (1.6 g/m²) coated from dichloromethane solvent.

(3) Overcoat layer of Fluorad FC-431 (a perfluorosulfonamido surfactant of 3M Corp.) (0.02 g/m²), 510 Silicone Fluid (a partial phenyl substituted polydimethylsiloxane of Dow Corning) (0.02 g/m²) in the linear condensation polymer described above (0.22 g/m²) coated from dichloromethane solvent.

On the reverse (back) side of these supports a subbing layer as described above was coated. On top of this layer, backing layers of this invention or control backing layers were each coated from water. The backing layers contained polyvinyl alcohol (fully hydrolyzed) from different suppliers, colloidal silica (LUDOX AM alumina modified colloidal silica of duPont) of approximately 0.014 μm diameter, and polystyrene beads crosslinked with m- and p-divinylbenzene of average diameter 4 μm . Polyethylene oxide was added as a binder and to control viscosity for coating. For coating ease, all backing layers also contained Triton X200E (a sulfonated aromatic-aliphatic surfactant of Rohm and Hass) and APG-225 (an alkyl glycoside surfactant of Henkel Corp.). Potassium chloride was added as an antistatic agent.

The following invention and control backing layers were prepared:

Invention Backing Layer E-7

Colloids 7190-25 (a fully hydrolyzed polyvinyl alcohol) (Colloids Industries)	0.065 g/m ²
Ludox AM	0.027 g/m ²
Polystyrene beads	0.003 g/m ²
Polyethyleneoxide #343 (a polyethylene oxide of mw 900,000) (Scientific Polymer Products)	0.019 g/m ²
Triton X200E	0.002 g/m ²
APG-225	0.002 g/m ²
Potassium Chloride	0.008 g/m ²

40 Invention Backing Layer E-8: As E-7 except Polyethyleneoxide #344 (Scientific Polymer Products) (mw 4,000,000)(0.019 g/m²) was used in place of Polyethyleneoxide #343.

Invention Backing Layer E-9:

Colloids 7190-25	0.27 g/m ²
Ludox AM	0.11 g/m ²
Polystyrene beads	0.003 g/m ²
Polyethyleneoxide # 136D (Scientific Polymer Products, a polyethylene oxide of mw 300,000)	0.065 g/m ²
Triton X200E	0.002 g/m ²
APG-225	0.002 g/m ²
Potassium Chloride	0.008 g/m ²

55 Invention Backing Layer E-10: As E-9 except 0.08 g/m² of Polyethyleneoxide #136D was used.

Invention Backing Layer E-11: As E-9 except 0.32 g/m² of Colloids 7190-25 was used.

Invention Backing Layer E-12: As E-11 except 0.13 g/m² Ludox AM was used.

EP 0 556 797 A1

Invention Backing Layer E-13: As E-12 except 0.38 g/m² of Colloids 7190-25 was used.

A control backing layer based on U.S. Pat. No. 5,011,814 that did not contain polyvinyl alcohol or large particles of polystyrene was also prepared on transparent polymeric film support:

Control Backing Layer C-3:	
Ludox AM	0.065 g/m ²
Polyethyleneoxide #136D	0.042 g/m ²
Triton X200E	0.002 g/m ²
APG-225	0.002 g/m ²
Potassium Chloride	0.008 g/m ²

A second control backing layer based on U.S. Pat. No. 5,096,875 containing large particles of polystyrene and small particles of colloidal silica but no polyvinyl alcohol was also prepared on transparent polymeric film support.

Control Backing Layer C-4:	
Ludox AM	0.065 g/m ²
Polystyrene beads	0.003 g/m ²
Polyethyleneoxide #136D	0.04 g/m ²
Triton X200E	0.002 g/m ²
APG-225	0.002 g/m ²
Potassium Chloride	0.008 g/m ²

Receiver backing layer to rubber pick roller friction and sliding friction between the backing layer of one receiver element and the receiving layer of an adjacent element were evaluated as described in Example 1. These values are indicated in Table II below.

Backside dye retransfer was evaluated as described in Example 1. The transferred Status A transmission densities were approximately 1.0 for each area. No background density correction was made for these transparency samples. The retransfer densities after one week 50°C, 50% RH are indicated in Table II below.

Table II

Receiver Element	Picking Friction (Newtons)	Sliding Friction (Newtons)	Retransferred Dye Density Status A-Transmission		
			Red	Green	Blue
C-3	7.1	7.6	0.11	0.17	0.14
C-4	7.5	4.7	0.06	0.07	0.06
E-7	7.7	5.3	0.03	0.03	0.03
E-8	7.9	4.4	0.02	0.03	0.02
E-9	8.6	4.8	0.03	0.03	0.03
E-10	8.3	4.1	0.04	0.04	0.04
E-11	8.0	3.9	0.04	0.04	0.03
E-12	7.5	3.6	0.03	0.04	0.04
E-13	7.7	3.0	0.04	0.05	0.04

All backing layers had coverages of less than 0.6 g/m², and produced receiver elements with good to excellent clarity. The data above show that the backing layers of the invention which contain polyvinyl alcohol have excellent high humidity picking friction and sliding friction characteristics compared to the prior

art control without large particles. In addition the invention backing layers with polyvinyl alcohol have significantly less backside dye retransfer compared to both controls.

Claims

5. 1. A dye-receiving element for thermal dye transfer comprising a support having on one side thereof a polymeric dye image-receiving layer and on the other side thereof a backing layer, characterized in that the backing layer comprises a mixture of polyvinyl alcohol as a polymeric binder, submicron colloidal inorganic particles, and polymeric particles of a size larger than the inorganic particles.
10. 2. The element of Claim 1, wherein the support comprises paper, and further comprising a polyolefin layer between the support and the backing layer.
15. 3. The element of claim 1, wherein the total coverage of the backing layer is from 0.1 to 2.5 g/m².
4. The element of Claim 1, wherein the backing layer further comprises polyethylene oxide as a polymeric binder in an amount by weight up to one half the total polymeric binder.
20. 5. The element of Claim 4, wherein the support is opaque and wherein the polyvinyl alcohol and polyethylenoxide are present in the backing layer at a total coverage of 0.1 to 0.4 g/m².
25. 6. The element of Claim 4, wherein the support is transparent and wherein the polyvinyl alcohol and polyethylene oxide are present in the backing layer in a ratio of at least 3:1 and a total coverage of 0.05 to 0.4 g/m².
7. The element of claim 1, wherein the support is transparent and the total coverage of the backing layer is from 0.1 to 0.6 g/m².
30. 8. The element of Claim 1, characterized in that the backing layer comprises a mixture of 10 to 80 wt.% polyvinyl alcohol as a polymeric binder, 0 to 15 wt.% polyethylene oxide as a polymeric binder, 15 to 80 wt.% submicron colloidal inorganic particles of a size from 0.01 to 0.05 μm, and 1 to 35 wt.% polymeric particles of a size from 1 to 15 μm, the polyvinyl alcohol comprising at least one half of the total amount of polymeric binder by weight.
35. 9. A process of forming a dye transfer image in a dye-receiving element comprising:
 - (a) removing an individual dye-receiving element comprising a support having on one side thereof a polymeric dye image-receiving layer and on the other side thereof a backing layer from a stack of dye-receiving elements;
 - (b) moving the individual dye-receiving element to a thermal printer printing station and into superposed relationship with a dye-donor element comprising a support having thereon a dye-containing layer so that the dye-containing layer of the donor element faces the dye image-receiving layer of the receiving element; and
 - (c) imagewise-heating the dye-donor element and thereby transferring a dye image to the individual dye-receiving element;
40. characterized in that the backing layer comprises a mixture of polyvinyl alcohol as a polymeric binder, submicron colloidal inorganic particles, and polymeric particles of a size larger than the inorganic particles.
45. 10. The process of claim 9, characterized in that the backing layer comprises a mixture of 10 to 80 wt.% polyvinyl alcohol as a polymeric binder, 0 to 15 wt.% polyethylene oxide as a polymeric binder, 15 to 80 wt.% submicron colloidal inorganic particles of a size from 0.01 to 0.05 μm, and 1 to 35 wt.% polymeric particles of a size from 1 to 15 μm, the polyvinyl alcohol comprising at least one half of the total amount of polymeric binder by weight.



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EUROPEAN SEARCH REPORT

Application Number

EP 93 10 2471

DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
A,D	EP-A-0 464 681 (EASTMAN KODAK COMPANY) * page 3, line 4 - line 30; claim 1 * * page 7, line 1 - line 50 * -----	1-10	B41M5/00
A,D	EP-A-0 444 588 (EASTMAN KODAK COMPANY) * page 2, line 36 - line 40; claim 1 * -----	1-10	
TECHNICAL FIELDS SEARCHED (Int. Cl.5)			
B41M			
The present search report has been drawn up for all claims			
Place of search	Date of compilation of the search	Examiner	
THE HAGUE	27 APRIL 1993	BACON A.J.	
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